

BLOQUE III:

TEMA 14.

ANABOLISMO DE GLÚCIDOS

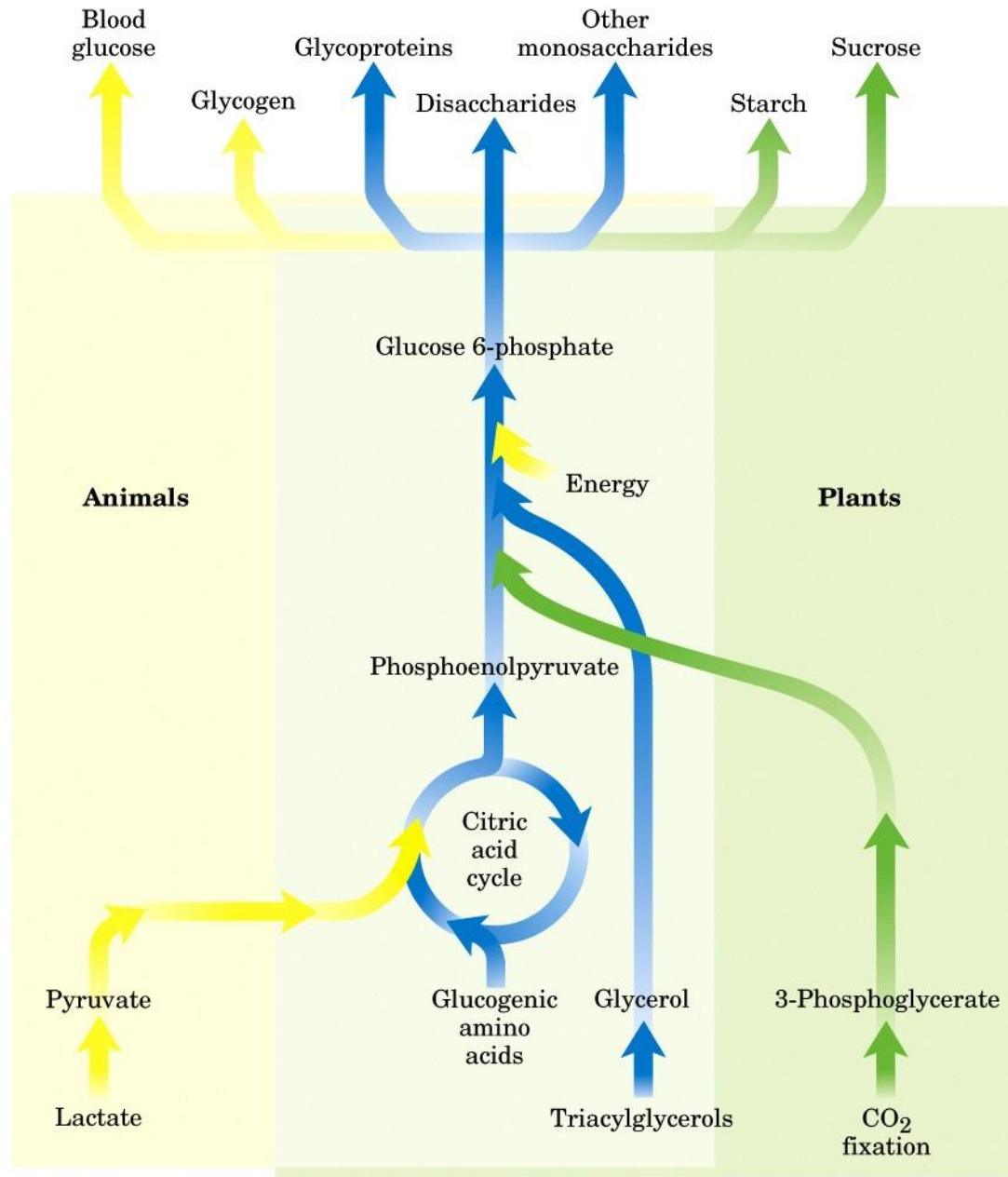
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ÍNDICE

- Gluconeogénesis
- Síntesis de glucógeno

Síntesis de glúcidos a partir de precursores sencillos



Los ácidos grasos no son precursores de la gluconeogénesis

GLUCONEOGÉNESIS

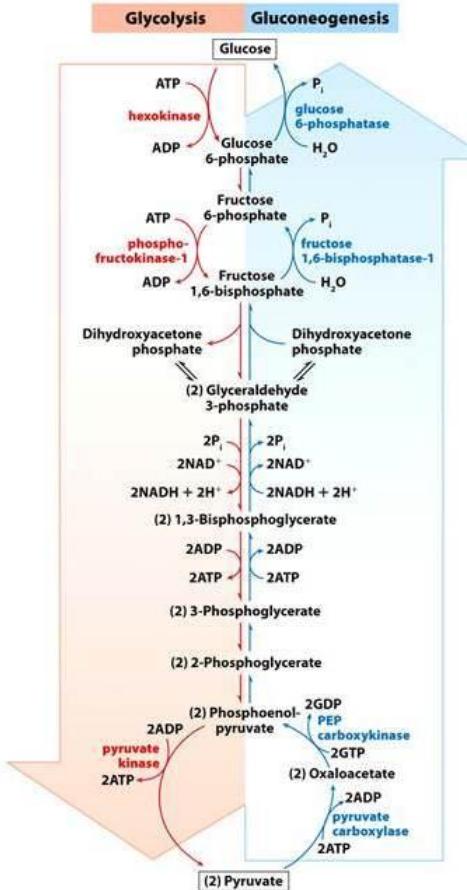
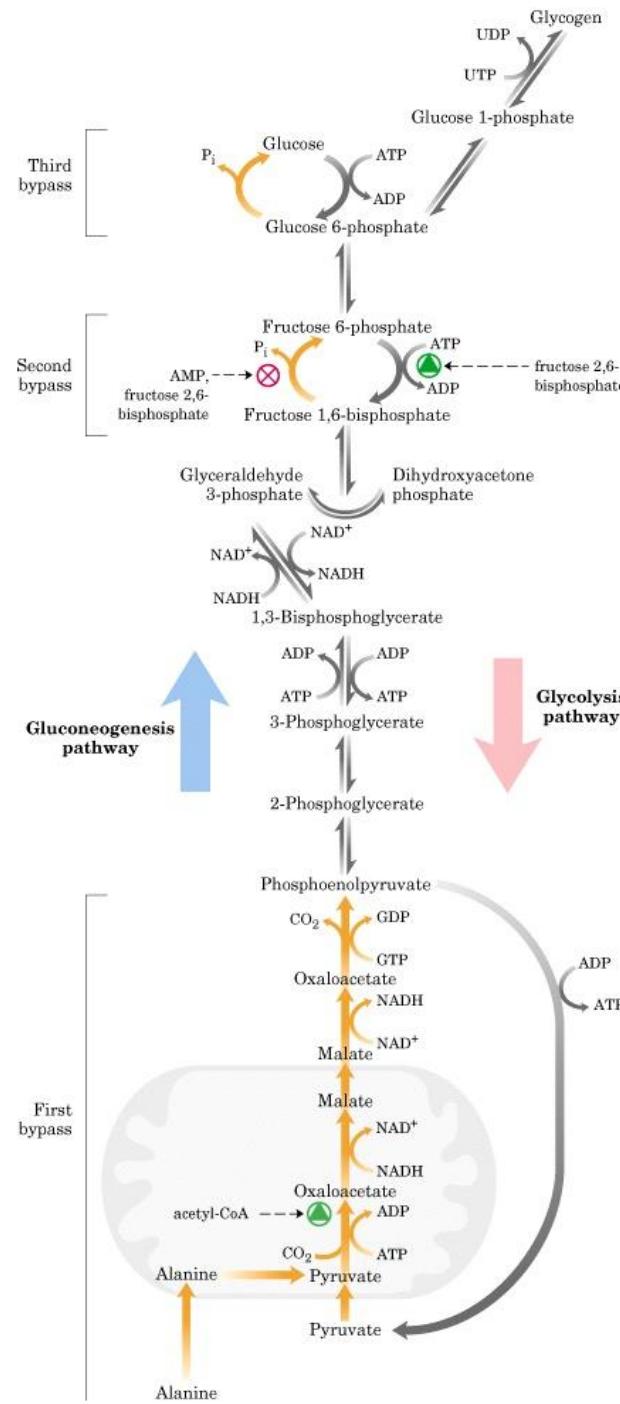
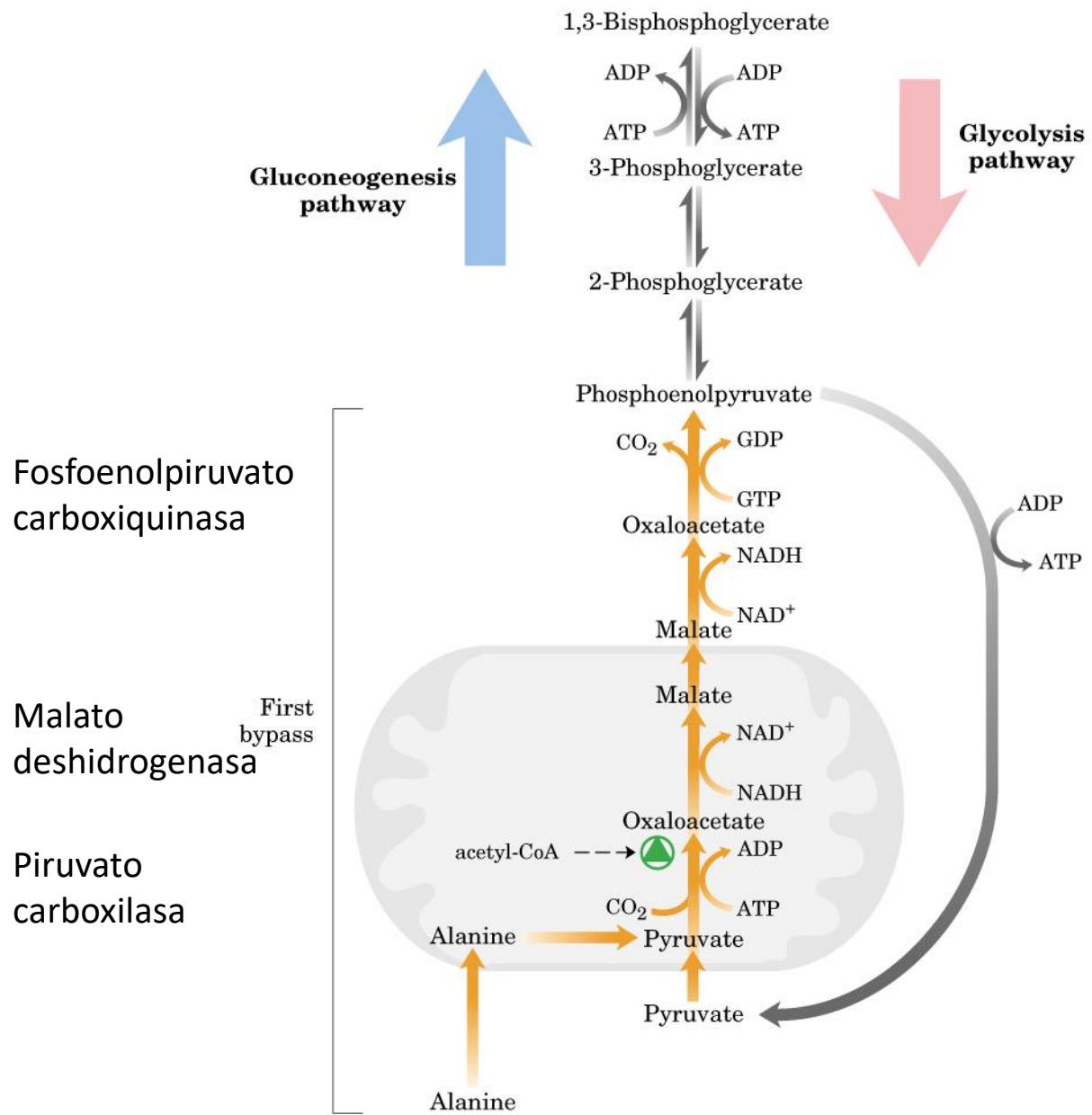
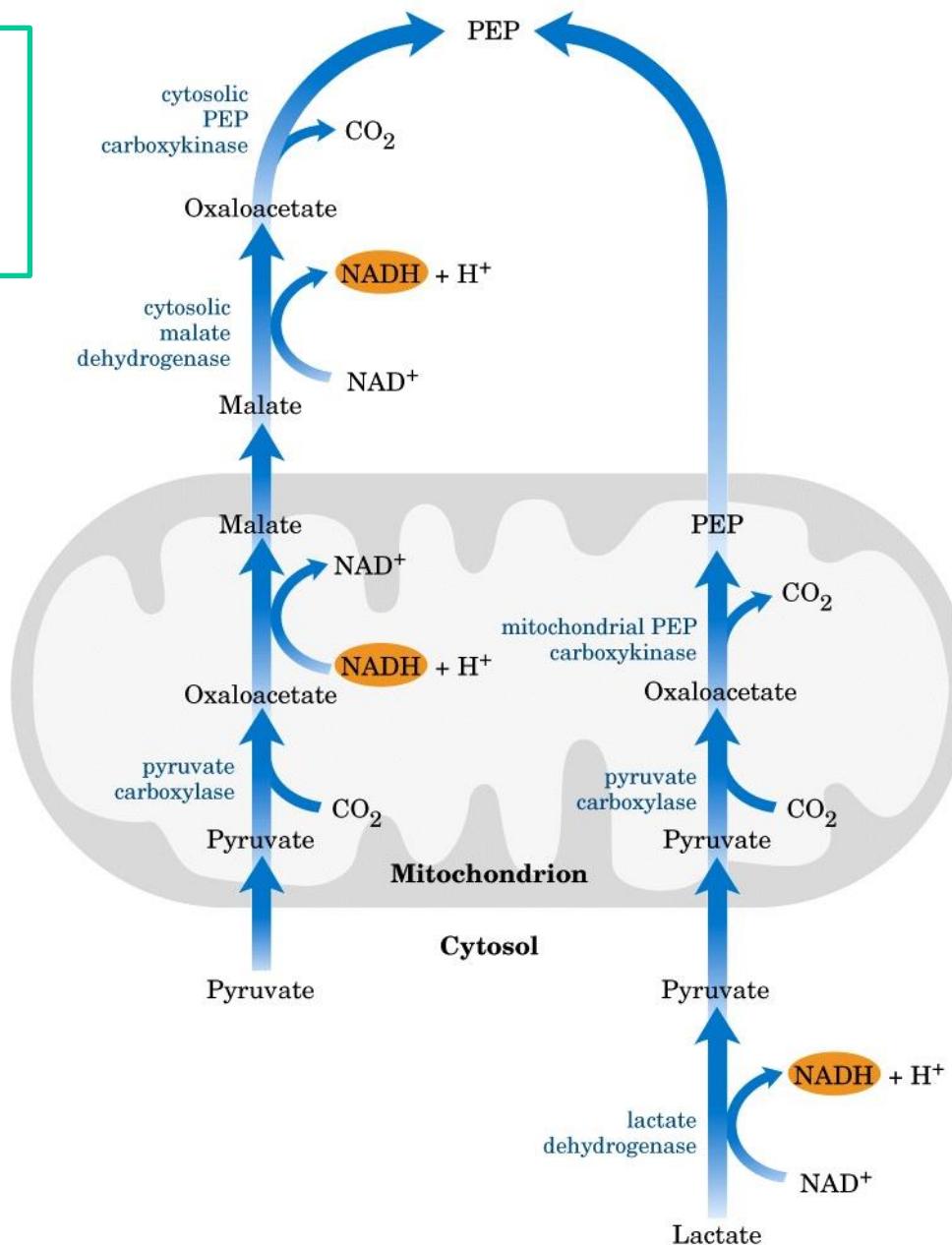


Figure 14-17
Lehninger Principles of Biochemistry, Sixth Edition
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Rutas alternativas desde el piruvato al fosfoenolpiruvato



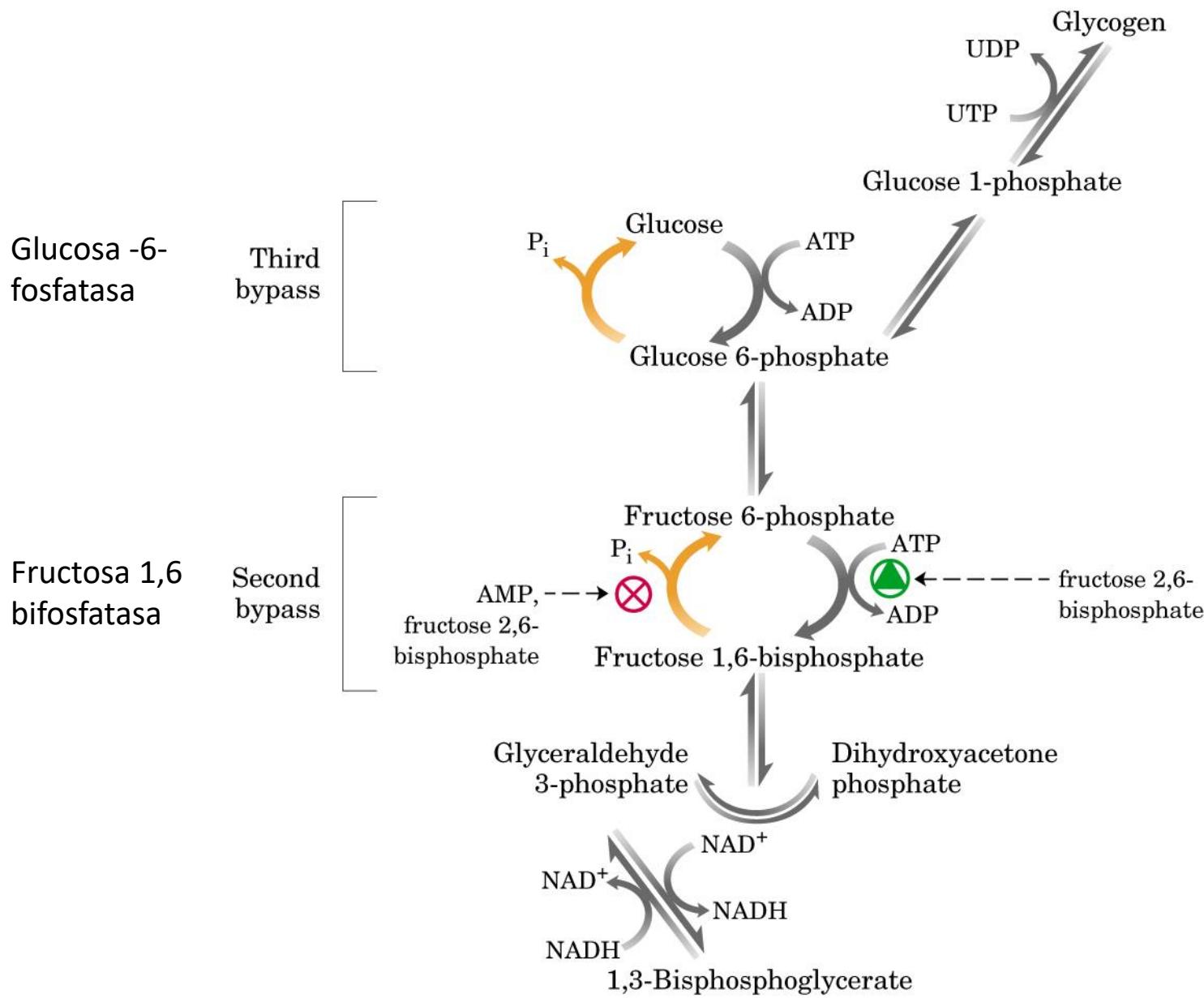


table 20–1

Free-Energy Changes of Glycolytic Reactions in Erythrocytes*

Glycolytic reaction step	ΔG° (kJ/mol)	ΔG (kJ/mol)
① Glucose + ATP \longrightarrow glucose 6-phosphate + ADP + H ⁺	-16.7	-33.4
② Glucose 6-phosphate \rightleftharpoons fructose 6-phosphate	1.7	-2.5
③ Fructose 6-phosphate + ATP \longrightarrow fructose 1,6-bisphosphate + ADP + H ⁺	-14.2	-22.2
④ Fructose 1,6-bisphosphate \rightleftharpoons dihydroxyacetone phosphate + glyceraldehyde 3-phosphate	23.8	-1.25
⑤ Dihydroxyacetone phosphate \rightleftharpoons glyceraldehyde 3-phosphate	7.5	2.5
⑥ Glyceraldehyde 3-phosphate + P _i + NAD ⁺ \longrightarrow 1,3-bisphosphoglycerate + NADH + H ⁺	6.3	-1.7
⑦ 1,3-Bisphosphoglycerate + ADP \rightleftharpoons 3-phosphoglycerate + ATP	-18.8	1.25
⑧ 3-Phosphoglycerate \rightleftharpoons 2-phosphoglycerate	4.4	0.8
⑨ 2-Phosphoglycerate \rightleftharpoons phosphoenolpyruvate + H ₂ O	7.5	-3.3
⑩ Phosphoenolpyruvate + ADP + H ⁺ \longrightarrow pyruvate + ATP	-31.4	-16.7

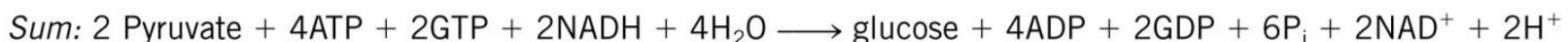
* ΔG° is the standard free-energy change, as defined in Chapter 14 (see p. 494). At pH 7.0, ΔG is the free-energy change calculated from the actual concentrations of glycolytic intermediates present under physiological conditions in erythrocytes. The glycolytic reactions bypassed in gluconeogenesis are shown in red.



table 20–2

Sequential Reactions in Gluconeogenesis Starting from Pyruvate*

Pyruvate + HCO ₃ ⁻ + ATP → oxaloacetate + ADP + P _i + H ⁺	x2
Oxaloacetate + GTP ⇌ phosphoenolpyruvate + CO ₂ + GDP	x2
Phosphoenolpyruvate + H ₂ O ⇌ 2-phosphoglycerate	x2
2-Phosphoglycerate ⇌ 3-phosphoglycerate	x2
3-Phosphoglycerate + ATP ⇌ 1,3-bisphosphoglycerate + ADP + H ⁺	x2
1,3-Bisphosphoglycerate + NADH + H ⁺ ⇌ glyceraldehyde 3-phosphate + NAD ⁺ + P _i	x2
Glyceraldehyde 3-phosphate ⇌ dihydroxyacetone phosphate	
Glyceraldehyde 3-phosphate + dihydroxyacetone phosphate ⇌ fructose 1,6-bisphosphate	
Fructose 1,6-bisphosphate + H ₂ O → fructose 6-phosphate + P _i	
Fructose 6-phosphate ⇌ glucose 6-phosphate	
Glucose 6-phosphate + H ₂ O → glucose + P _i	



*The bypass reactions are in red; all other reactions are reversible steps of glycolysis. The figures at the right indicate that the reaction is to be counted twice, because two three-carbon precursors are required to make a molecule of glucose. Note that the reactions required to replace the cytosolic

NADH consumed in the glyceraldehyde 3-phosphate dehydrogenase reaction (the conversion of lactate to pyruvate in the cytosol or the transport of reducing equivalents from mitochondria to the cytosol in the form of malate) are not considered in this summary.



table 20–3

**Glucogenic Amino Acids, Grouped
by Site of Entry***

Pyruvate

Alanine

Cysteine

Glycine

Serine

Tryptophan[†]

α -Ketoglutarate

Arginine

Glutamate

Glutamine

Histidine

Proline

Succinyl-CoA

Isoleucine[†]

Methionine

Threonine

Valine

Fumarate

Phenylalanine[†]

Tyrosine[†]

Oxaloacetate

Asparagine

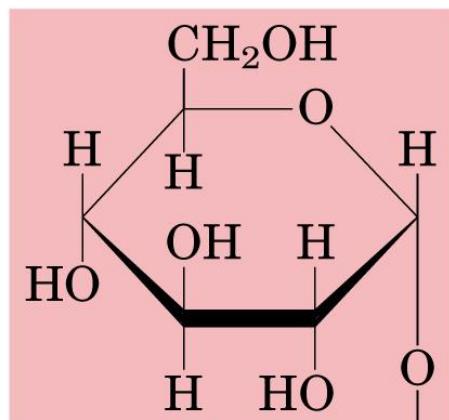
Aspartate

*These amino acids are precursors of blood glucose or liver glycogen because they can be converted to pyruvate or citric acid cycle intermediates. Only leucine and lysine are unable to furnish carbon for net glucose synthesis.

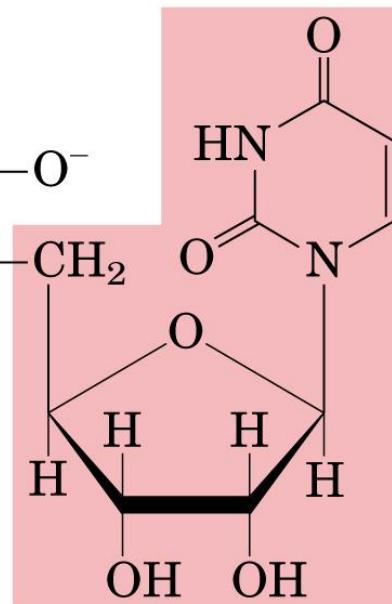
[†]These amino acids are also ketogenic (see Fig. 18–19).

BIOSÍNTESIS DE GLUCÓGENO

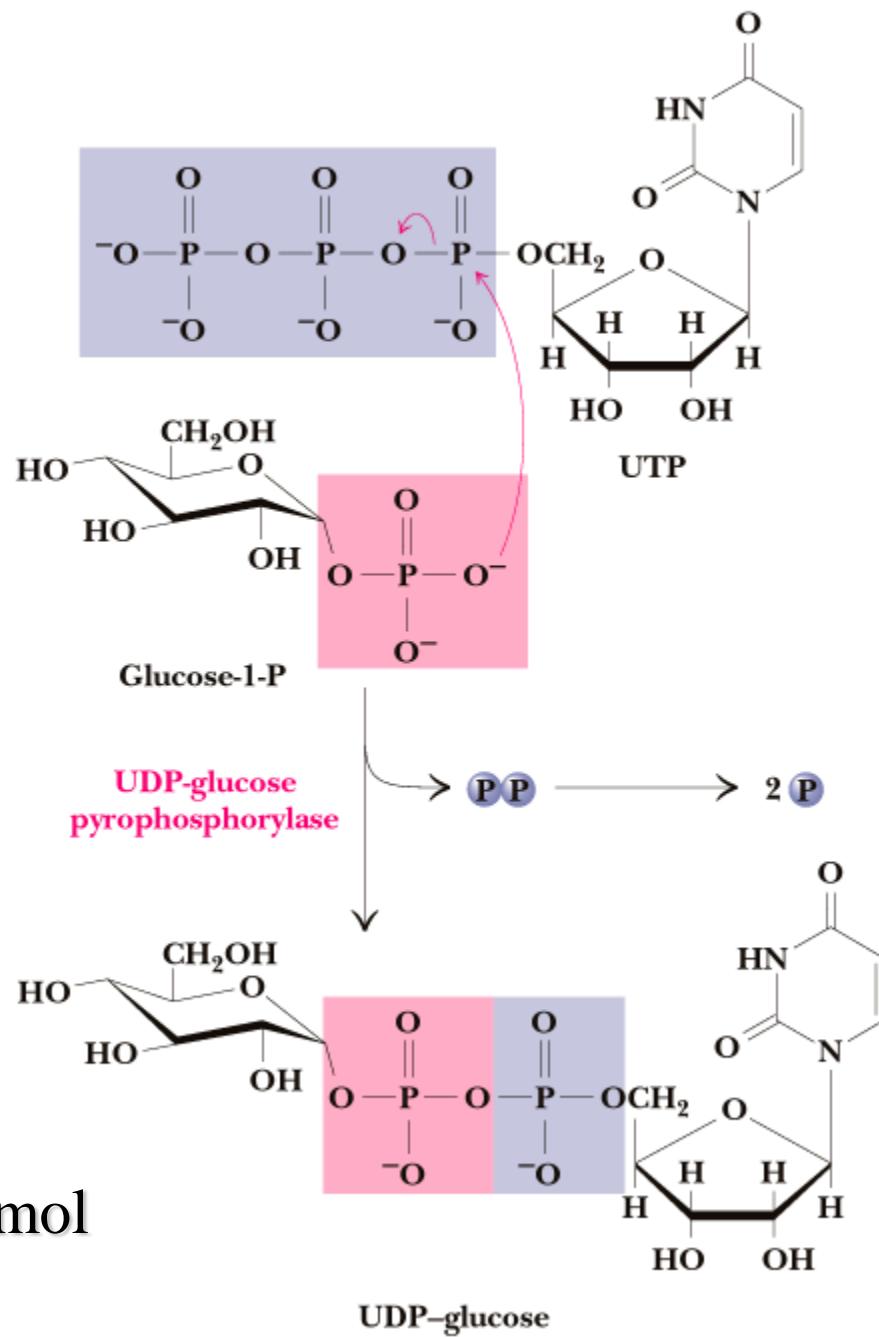
D-Glucosyl group

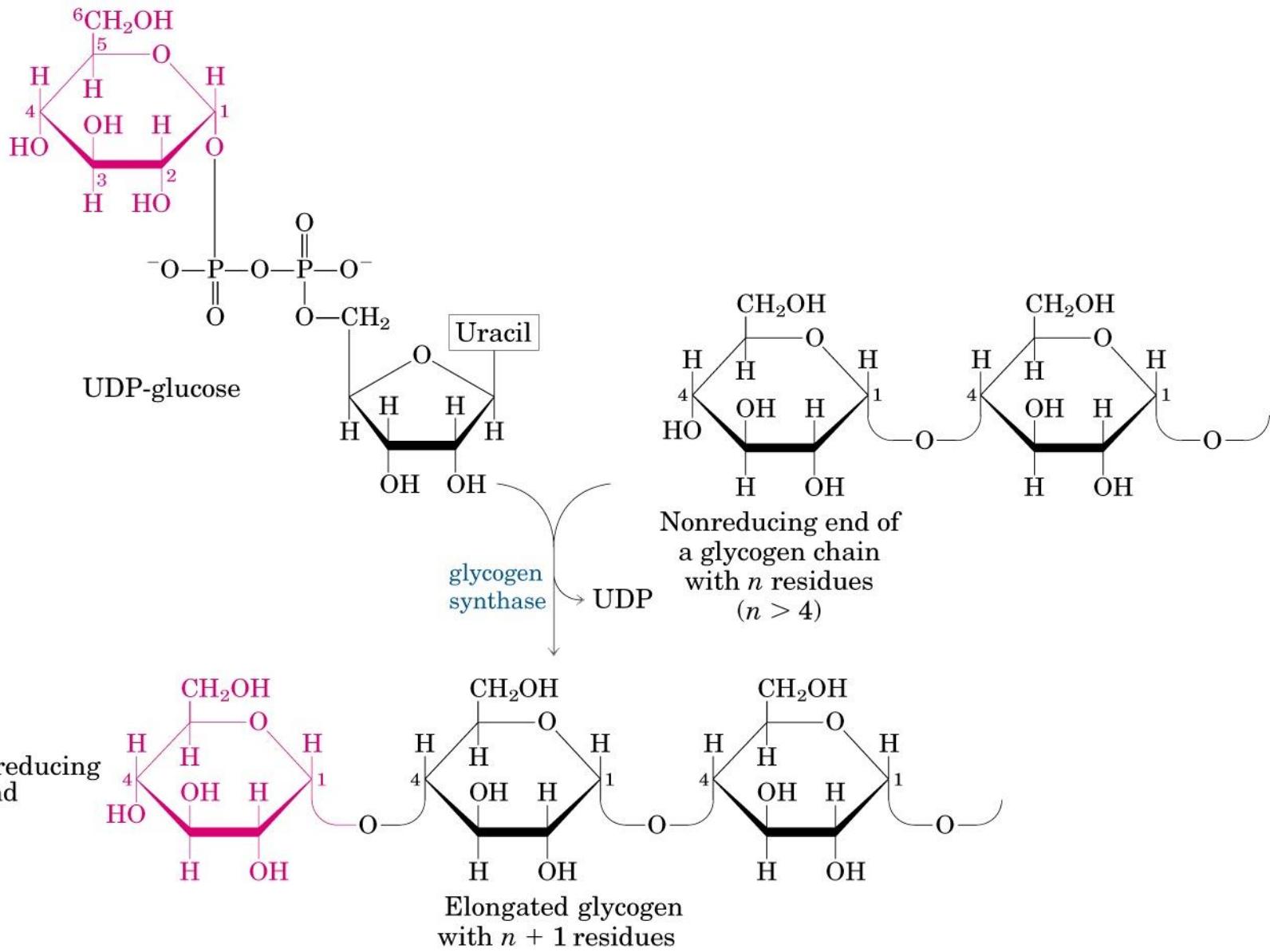


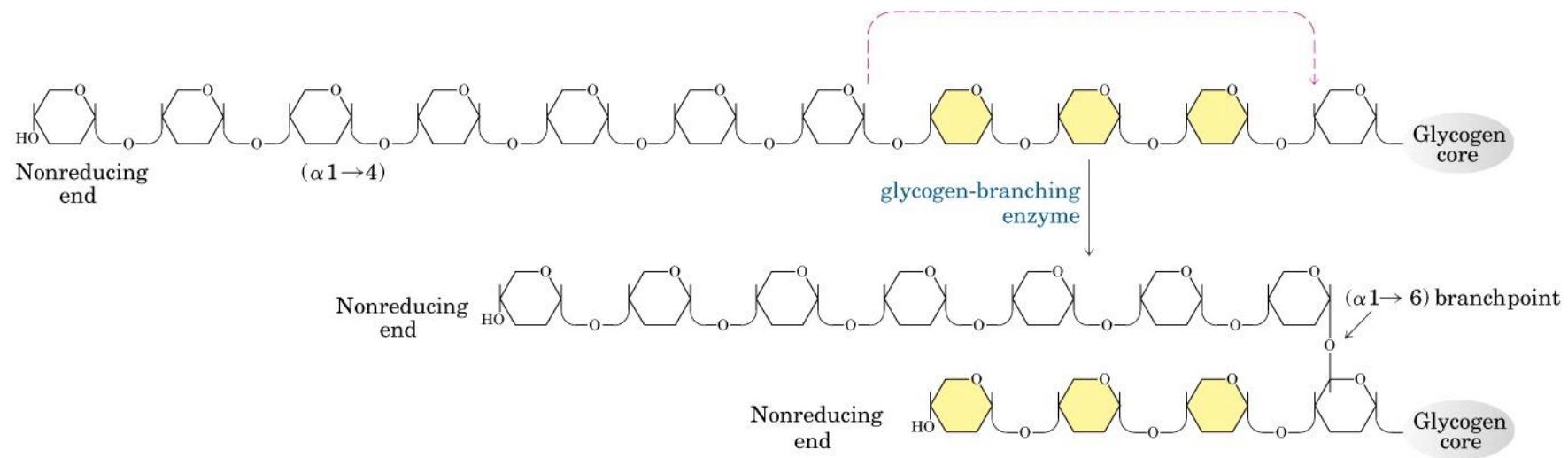
Uridine



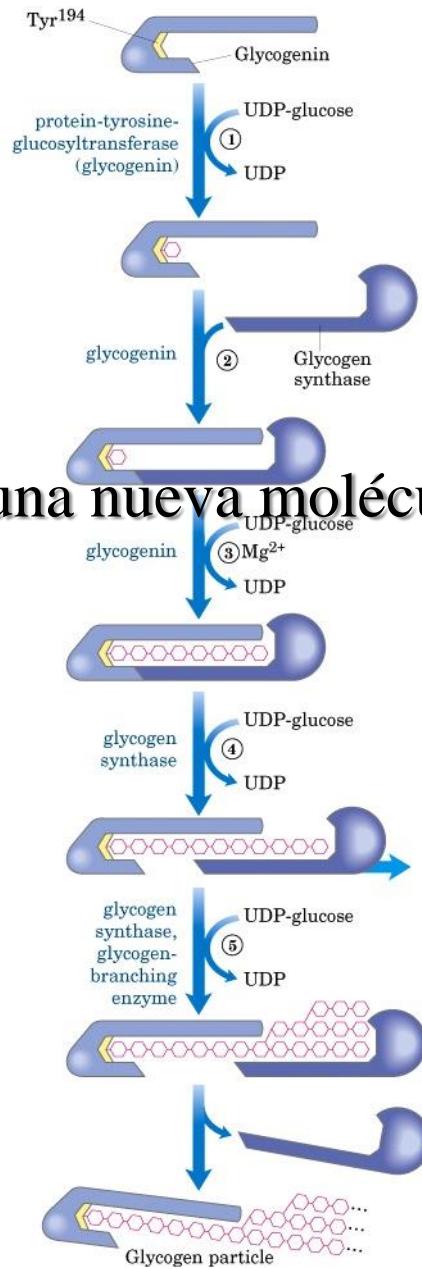
UDP-glucose



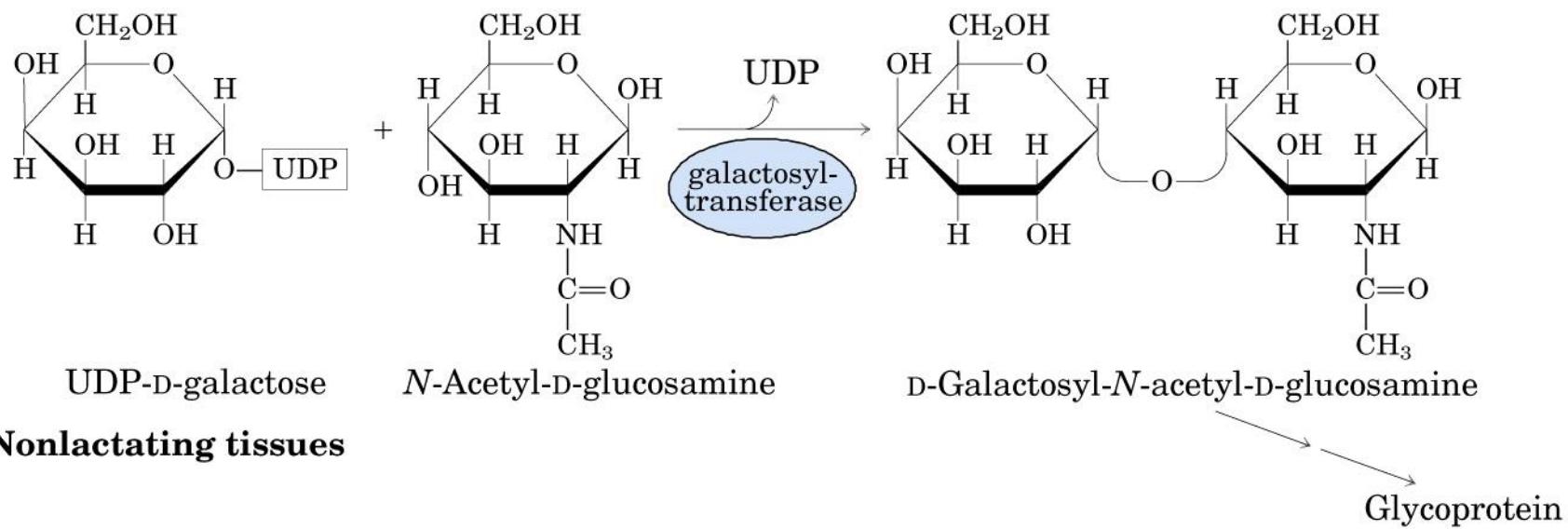




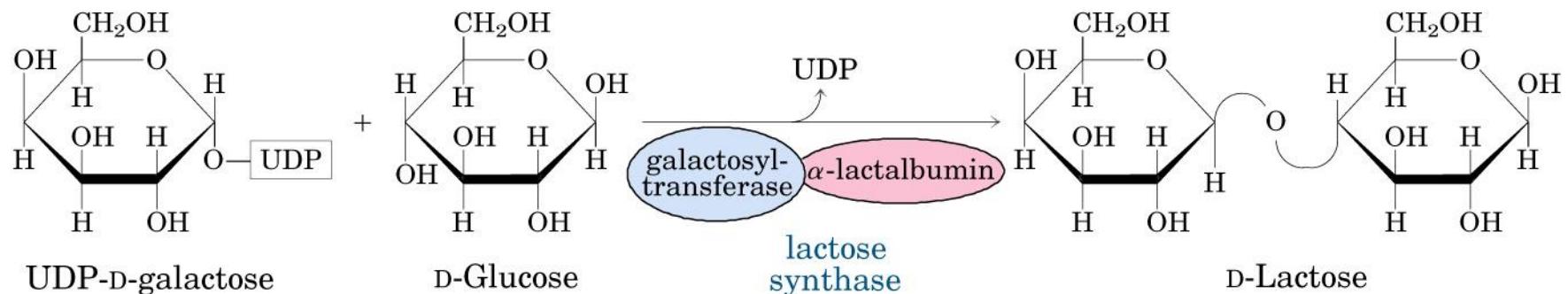
¿Cómo se inicia una nueva molécula de glucógeno?



Síntesis de lactosa



(a) Nonlactating tissues



(b) Lactating mammary gland

Síntesis de lactosa

